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 (a) $\int \dots \int_{0 \leq \sum_{i=1}^{3N} |x_i| \leq R} \prod_{i=1}^{3N} dx_i \equiv V$ and let $\sum_{i=1}^{3N} |x_i| = R \Rightarrow \prod_{i=1}^{3N} dx_i = dR$

Thus, per C.3 $dV = (3N) C_{3N} R^{3N-1} dR \Rightarrow \int_{-\infty}^{\infty} dV = 3N C_{3N} \int_0^{\infty} dR R^{3N-1}$

→ Integration trick! multiply both sides by e^{-R} .

$\Rightarrow \int_{-\infty}^{\infty} dV e^{-R} = 3N C_{3N} \int_0^{\infty} dR e^{-R} R^{3N-1}$

↑
The radial part

→ To evaluate the integral, we once again note that $\int_0^{\infty} e^{-x} x^k dx = k!$

$\Rightarrow \int_{-\infty}^{\infty} dV e^{-R} = 3N C_{3N} (3N-1)!$

$\int_0^{\infty} \prod_{i=1}^{3N} dx_i e^{-\sum_{i=1}^{3N} |x_i|} = 3N C_{3N} (3N-1)!$

$\left[\int_0^{\infty} dx_i e^{-|x_i|} \right]^{3N} = 3N C_{3N} (3N-1)! \Rightarrow 8\pi = 3N C_{3N} (3N-1)!$

$\therefore C_{3N} = \frac{2^{3N}}{(3N)!}$ Thus $V_{3N} = \frac{(2R)^{3N}}{(3N)!}$ by C.2

Now, $p = \hbar k = \hbar \frac{m v}{D} \Rightarrow E = \frac{c \hbar m v}{D} \Rightarrow E = \frac{c \hbar \pi}{D} \sum_{i=1}^{3N} n_i \Rightarrow R = \frac{ED}{c \hbar \pi}$

The "volume in phase space" is $\omega =$

$\omega = V(E/c) \cdot D^{3N} = \frac{(2ED)^{3N}}{c^{3N} (3N)!} = \omega$

$\int \dots \int_{0 \leq \sum_{i=1}^{3N} |p_i| \leq E/c} d^{3N} p \cdot \int \dots \int d^{3N} z$
 $\hookrightarrow D^{3N}$

where the 1-D box is of length D.

Now we look for Ω : the number of ways of distributing E amongst the particles. As explained in 1-8 and 2-7,

$\Omega = \frac{(R+N-1)!}{R! (N-1)!} \xrightarrow{\text{see 2-7}} \Omega \approx \frac{(ED)^{3N}}{c^{3N} (3N)!} \frac{1}{(\hbar \pi)^{3N}} = \frac{(ED)^{3N}}{c^{3N} (3N)!} \left(\frac{2}{h}\right)^{3N} = \Omega$

$\Omega = \frac{\omega}{\omega_0} \Rightarrow \omega_0 = \frac{\Omega}{\omega} = \frac{(2ED)^{3N}}{c^{3N} (3N)!} \frac{1}{\left(\frac{2ED}{h}\right)^{3N}} = \boxed{h^{3N}}$



(b) Now I use the Ω calculated in (a) to find $\frac{C_P}{C_V}$, etc.: $\Omega = \frac{(2E)^{3N}}{(hc)^{3N} (3N)!}$

$$S = k \ln \Omega = k \ln \left[\frac{(2E)^{3N}}{(hc)^{3N} (3N)!} \right]$$

$$\Rightarrow T = \frac{1}{\left(\frac{\partial S}{\partial E}\right)_{N,V}} = \frac{E}{3kN} = T \quad \leftarrow \text{(see mathematical)}$$

Setting $D^{3N} = V^N$ $\{V^N = D^N; \downarrow \text{I-D particles}\}$
 $V^N = D^N$

$$P = \frac{\left(\frac{\partial S}{\partial V}\right)_{N,E}}{\left(\frac{\partial S}{\partial E}\right)_{N,V}} = \frac{E}{3V} = P \quad \Rightarrow \quad PV = \frac{E}{3}$$

We see from the expression for T that $E = 3kNT$,

so $C_V = \left(\frac{\partial E}{\partial T}\right)_{N,V} = 3Nk = C_V$

and $C_P = \left(\frac{\partial(E+PV)}{\partial T}\right)_{N,P} = \left(\frac{\partial(E + \frac{E}{3})}{\partial T}\right)_{N,P} = \frac{\partial 4NkT}{\partial T} = 4Nk = C_P$

$$\Rightarrow \frac{C_P}{C_V} = \frac{4}{3}$$

Also $\mu = \left(\frac{\partial E}{\partial N}\right)_{V,S} = 3kT = \mu$

Assuming we set $D^{3N} = V^N$, it all works out the same!

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$$S[N, V, E] := k \log \left[\frac{V^N (2E)^{3N}}{(hc)^{3N} (3N)!} \right]$$

$$\frac{1}{D[S[N, V, E], E]} \quad (*\text{Temperature}*)$$

$$\frac{E}{3kN}$$

$$\frac{D[S[N, V, E], V]}{D[S[N, V, E], E]} \quad (*\text{Pressure}*)$$

$$\frac{E}{3V}$$